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Publisher *Taylor & Francis*

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Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713617200>

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Claude Monneret^a; Roselyne Gagnet^a; Jean-Claude Florent^a

^a Département de Pharmacognosie associé au CNRS, UA 484, Université Rene Descartes, Paris cédex 06

To cite this Article Monneret, Claude , Gagnet, Roselyne and Florent, Jean-Claude(1987) 'Regioselective Alkylation and Acylation of Methyl, 2,6-Dideoxyhexo-Pyranosides Via Their Stannylene Acetals: Key Step for the Synthesis of Daunosamine and Related Amino-Sugars', *Journal of Carbohydrate Chemistry*, 6: 2, 221 – 229

To link to this Article: DOI: 10.1080/07328308708058872

URL: <http://dx.doi.org/10.1080/07328308708058872>

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REGIOSELECTIVE ALKYLATION AND ACYLATION OF METHYL 2,6-DIDEOXYHEXOPYRANOSIDES VIA THEIR STANNYLENE ACETALS: KEY STEP FOR THE SYNTHESIS OF DAUNOSAMINE AND RELATED AMINO-SUGARS

Claude Monneret^{*}, Roselyne Gagnet and Jean-Claude Florent

Département de Pharmacognosie associé au CNRS, UA 484, Université René Descartes, 4 avenue de l'Observatoire, 75270 Paris cédex 06.

Received September 22, 1986 - Final Form November 7, 1986

ABSTRACT

Synthetic intermediates for the preparation of 3-amino-2,3,6 (or 4-amino-2,4,6)-trideoxy-L-hexopyranosides, sugar moieties of anthracycline antibiotics, were selectively obtained from stannylene acetals of methyl 2,6-dideoxy- α -L-lyxo and arabino-hexopyranosides.

INTRODUCTION

The synthesis of new antitumour anthracyclines¹, analogs of daunorubicin and doxorubicin modified in their sugar moiety, and of oligosaccharides², natural constituents of class II anthracyclines, required in both cases selective acylation or alkylation of the 3-OH of methyl-2,6-dideoxy-L-hexopyranosides. This led us to synthesize for example 4-amino-2,4,6-trideoxy-L-hexopyranosides³ from the corresponding L-arabino isomer (methyl 2-deoxy-L-rhamnopyranoside), while elaboration of the oligosaccharide of Aclacinomycin A was based upon selective O-benzylation of the corresponding L-lyxo-isomer (methyl-2-deoxy-L-fucoside)⁴.

On the other hand, utility of organotin derivatives of alcohols in regioselective acylation, alkylation or oxidation reactions has been largely demonstrated⁵ and the synthetic application of these stannylene acetals in the case of polyols such as carbohydrates is well documented.⁶ The observed regioselectivity of those reactions seems to be usually determined by steric factors as demonstrated by substitution at the equatorial oxygen of vicinal axial-equatorial dibutylstannylene derivatized hydroxy groups.⁷

RESULTS AND DISCUSSION

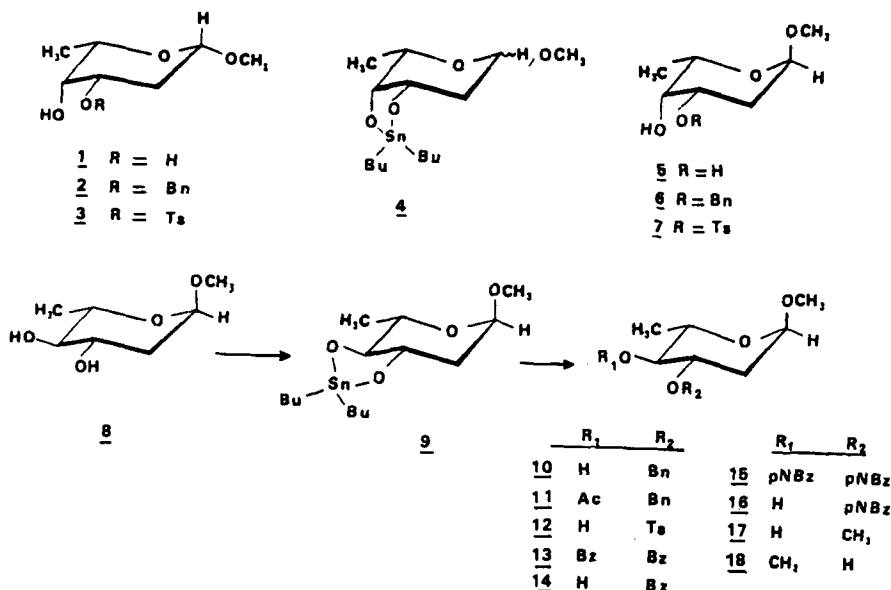
We previously reported⁴ that treatment of methyl 2,6-dideoxy- β -L-lyxo-hexopyranoside (1) with dibutyltin oxide in benzene followed by alkylation (BnBr) catalyzed by tetrabutylammonium iodide⁸ led regiospecifically, as might be expected, to the 3-O-benzyl compound 2 in 95 % yield after purification (lack of C-2 substituent and equatorial orientation of the 3 C-O bond).

The same regiospecificity was again observed when the dibutyl stannylene derivatives⁹ of methyl 2,6-dideoxy- β - and α -L-lyxo-hexopyranoside (1) and (5) (i.e. 4 on scheme) were treated with TsCl for 24 h at room temperature giving 3 and 7 in 90 and 98 % yields respectively after chromatography (dichloromethane-MeOH, 98:2). On the other hand, benzylation of the stannylene acetal of 5 gave regioselectively, as already reported¹⁰, the 3-O-benzyl derivative 6 in 95 % yield after chromatography (hexane-EtOAc, 4:1). Consequently, the regioselectivity of these reactions is not influenced by the anomeric configuration.

The next question was : could any selectivity be observed during alkylation or acylation of the stannylene acetal of methyl 2,6-dideoxy- α -L-arabino-hexopyranoside (8) (e.g. 9 on scheme) where the two vicinal OH groups at C-3 and C-4 are trans diequatorial in the ¹C₄ conformation of 8 ? If, as generally assumed, such regioselectivity depends on steric factors (or stereoelectronic effects), the lack of substituent at C-2 could be in favour of reaction at O-3. Actually, a noteworthy regioselectivity was observed when 9

was treated with BnBr under reflux in the presence of tetrabutylammonium iodide giving 10 in 70 % yield after chromatography (hexane-EtOAc, 7:3). Structural determination of 10 was based upon ^1H NMR data. The upfield shift of 4-H (dd at 3.19) in 10 with respect to the corresponding acetyl derivative 11 (dd at 4.77) was in agreement with the deshielding influence of esterification of the 4-OH.

Interestingly, when the stannylene complex 9 was stirred for 24 h at room temperature with TsCl, the 3-O-tosyl derivatives 12¹¹ was regioselectively obtained in 70 % yield after chromatography (dichloromethane-MeOH, 98:2).



Although a little amount of 3,4-di-O-benzoyl derivative 13¹² ($\approx 5\%$) was obtained when 9 in toluene solution was stirred for 2 h at -45°C with benzoyl chloride (1.1 molar) and triethylamine (1.1 molar eq.), the 3-O-benzoyl derivative 14¹² was regioselectively obtained in 70 % yield after chromatography (hexane-EtOAc, 4:1). A similar result was obtained when p-nitrobenzoyl chloride was used instead of benzoyl chloride, 15 and 16 being isolated in 5 and 70 % yields respectively.

Compared with acylation and tosylation, alkylation of the tin intermediate 9 with methyl iodide required a large excess of alkylating agent, a longer reaction time and a higher temperature even in the presence of tetrabutylammonium iodide. Chromatography of the crude material with hexane-acetone (3:1) as the eluent afforded successively the 3-O-methyl derivative, methyl α -L-oleandroside¹³ (17) and the 4-O-methyl derivative 18 in 52 % and 18 % yields respectively.

In conclusion selective preparation of 3-O-derivatives of methyl 2,6-dideoxy- α -L-lyxo and α -L-arabino hexopyranosides illustrates an extremely useful application of organotin alkoxides in carbohydrate chemistry. The orientation of attack could be due to the 3-oxygen atom preferentially binding apically in the dimeric structure which probably exists in benzene¹⁴. However, as regioselective tosylation¹¹ and acylation¹² of the 3-oxygen atom were already reported when the diol derivative 8 was treated with an equimolecular amount of appropriate reagent in pyridine at low temperature, both steric and electronic factors may play a deciding role and explain the present results.

According to this approach, several partially alkylated or acylated monosaccharides of high synthetic potential have been prepared in good yield : compounds 3 which has been previously used as a precursor of methyl 3-amino-2,3,6-trideoxy-L-xylo-hexopyranoside or 3-epi-L-daunosamine,¹⁰ and 12, which has been widely synthesized as a key intermediate in the preparation of L-daunosamine,¹¹ L-acosamine,¹⁵ L-ristosamine¹⁶ and L-megosamine.¹⁷ Other derivatives have been involved in syntheses of 4-amino-2,4,6-trideoxy-L-hexopyranoses (14 as precursor of the L-lyxo isomer¹² and 17 as precursor of the 3-O-methyl derivative of the L-arabino isomer or L-holantosamine¹³) or of branched-chain sugars (6 as precursor of L-rubranitrose¹⁰).

EXPERIMENTAL

General Methods and Material. Melting points were determined on a Hofler hot-stage microscope and are uncorrected. IR spectra

were recorded on a Perkin-Elmer Model 257 spectrophotometer, calibrated against polystyrene film and are expressed in cm^{-1} . ^1H NMR spectra were obtained on a Bruker HX 270 in CDCl_3 (s, singlet ; d, doublet ; m ; multiplet ; q, quadruplet) (TMS refer. 0 and coupling constants in Hertz). Silica gel for column chromatography or flash chromatography was Merck silica gel H.60 n°7736. Microanalyses were performed by the "Laboratoire de Microanalyse du CNRS" Gif-sur-Yvette.

General procedures for benzylation, tosylation and methylation.

A mixture of hexopyranoside (1 mmol) and dibutyltin oxide (1.2 mmol) in benzene (50 ml) was refluxed for 4-6 h with continuous removal of water. After concentration to half-volume, tetrabutylammonium iodide (0.5 mmol) was added and the following procedures used : Procedure A : BnBr (1.5 mmol) was added and the mixture stirred under reflux for 18-24 h ; Procedure B : TsCl (1.5 mmol) was added and the mixture was stirred at room temperature for 6-8 h ; Procedure C : MeI (15 mmol) was added in several portions while the mixture was stirred under reflux for 72 h.

General procedure for benzylation and p-nitrobenzylation, Procedure D : A mixture of hexopyranoside (1 mmol) and dibutyltin oxide (1.2 mmol) in toluene (50 ml) was refluxed for 4 h with continuous removal of water and then concentrated to half volume. After cooling at -45°C , Et_3N (1.1 mmol) and p-NBzCl or BzCl (1.1 mmol) were added. The mixture was stirred at the same temperature for 2 h and then, methanol (2-5 ml) was added before the reaction mixture was allowed to reach room temperature.

General work-up. Silica gel (≈ 1 g/mmol of starting material) was added to the reaction mixture and the solvent removed in vacuo. The resulting powder was flash chromatographed with eluent as indicated for each compound.

Methyl 2,6-dideoxy-3-O-tosyl- β -L-lyxo-hexopyranoside (3). Treatment of methyl 2,6-dideoxy- β -L-lyxo-hexopyranoside (1) (290 mg) according to procedure B afforded 550 mg (90 %) of 3 after flash chromato-

graphy (dichloromethane-methanol, 98:2). Constants and spectroscopic data of 3 were in full agreement with those previously reported.⁹

Methyl 3-O-benzyl-2,6-dideoxy- α -L-lyxo-hexopyranoside (6). Treatment of methyl 2,6-dideoxy- α -L-lyxo-hexopyranoside (5) (500 mg) following procedure A led to 500 mg (95 %) of 5 after flash chromatography (hexane-ethyl acetate, 4:1) : syrup, $[\alpha]_D^{20}$ -125° (c 1.8, chloroform) ; lit.¹⁰ : syrup, $[\alpha]_D^{20}$ -127° (c 1, chloroform).

Methyl 2,6-dideoxy-3-O-tosyl- α -L-lyxo-hexopyranoside (7). Treatment of 5 (300 mg) according to procedure B afforded 585 mg (98 %) of 7 after flash chromatography (hexane-ethyl acetate 2:1) : mp 95-96° (hexane-acetone 1:1), $[\alpha]_D^{20}$ -140° (c 1.1, chloroform) ; IR_{max}^{film} : 3420(OH) and 1600 cm⁻¹ (Ar). ¹H NMR : δ 7.72(d, J = 6) and 7.26 (d, J = 6)(AB, Aromat.), 4.82 (m, 3-H), 4.70 (d, J = 3, 1-H), 3.82 (q, J = 7, 5-H) ; 3.75 (broad s, 4-H), 3.25 (s, OCH₃), 2.44 (s, tosyl), 2.09 (m, J = 13, J' = 9, J'' = 3, 2a-H) and 1.70 (m, J = 13, J' = 5, 2e-H), 1.25 (d, J = 7, 6-CH₃).

Anal. Calcd for C₁₄H₂₀O₆S (316.31) : C, 53.15, H, 6.37, O, 30.34, S, 10.11 ; Found : C, 53.28, H, 6.45, O, 30.20, S, 10.02.

Methyl 3-O-benzyl-2,6-dideoxy- α -L-arabino-hexopyranoside (10). Treatment of 8 (1.6 g, 10 mmol) according to procedure A gave 1.75 g (70 %) after flash chromatography (hexane-ethyl acetate, 7:3) : syrup, $[\alpha]_D^{20}$ -40° (c 1.2, chloroform) ; IR_{max}^{film} 3440(OH) and 1500 cm⁻¹ (Ar) ; ¹H NMR : δ 4.73 (d, J = 4, 1-H), 3.74-3.59 (m, 3-H, and 5-H), 4.62 (d, J = 12) and 4.44 (d, J = 12) (CH₂ benzyl), 3.30 (s, OCH₃), 3.19 (dd, J = J' = 10, 4-H), 2.27 (dd, J = 13, J' = 4, 2e-H) and 1.61 (m, J = 13, J' = 10, J'' = 3, 2a-H), 1.29 (d, J = 7, 6-CH₃).

Anal. Calcd for C₁₄H₂₀O₄ (252.30) : C, 66.64, H, 7.99, O, 25.37 ; Found : C, 66.75, H, 8.03, O, 25.50.

Methyl 4-O-acetyl-3-O-benzyl-2,6-dideoxy- α -L-arabino-hexopyranoside (11). To a solution of 10 (150 mg, 0.6 mmol) in pyridine (15 ml) was added acetic anhydride (3 ml). After stirring at r.t. for 2 h, extraction with ether afforded 160 mg (89 %) of 11: syrup, $[\alpha]_D^{20} -84^\circ$ (c 1, chloroform); IR_{max}^{film}: 1740, 1230 and 1040 cm^{-1} (OAc); $^1\text{H NMR}$: δ 4.77 (d, J = 4, 1-H), 4.77 (dd, J = J' = 10, 4-H), 4.61 (d, J = 12) and 4.48 (d, J = 12) (CH₂ benzyl), 3.90-3.68 (m, 3-H and 5-H), 3.27 (s, OCH₃), 2.24 (dd, J = 13, J' = 5, 2e-H), 2.02 (s, OAc), 1.67 (m, J = 13, J' = 12, J'' = 4, 2a-H), 1.14 (d, J = 7, 6-CH₃).

Anal. Calcd for C₁₆H₂₂O₅ (294.35): C, 65.28, H, 7.54, O, 27.17; Found: C, 65.35, H, 7.60, O, 27.30.

Methyl 2,6-dideoxy-3-O-tosyl- α -L-arabino-hexopyranoside (12). Treatment of 8 (10 g, 62 mmol) according to procedure B gave 14.3 g (75 %) of 12 after flash chromatography (hexane-ethyl acetate, 3:1). M.p., IR and spectroscopic data were identical in all respects with literature data.^{11,17}

Methyl 3,4-di-O-benzoyl-2,6-dideoxy- and Methyl 3-O-benzoyl-2,6-dideoxy- α -L-arabino-hexopyranoside (13 and 14). Treatment of 8 (1 g, 6.16 mmol) according to procedure D with benzoyl chloride led to 78 mg (5 %) of 13 and 1.15 g (70 %) of 14 separated by flash chromatography (hexane-ethyl acetate, 4:1). 13 and 14 were identical in all respects with the literature data.¹²

Methyl 2,6-dideoxy-3,4-di-O-p-nitrobenzoyl and methyl 2,6-dideoxy 3-O-p-nitrobenzoyl- α -L-arabino-hexopyranoside (15 and 16). Treatment of 8 (1g, 6.16 mmol) according to procedure D (with p-nitrobenzoyl chloride) afforded 95 mg (5 %) of 15 and 1.98 g (70 %) of 16 easily separated by flash chromatography (hexane-ethyl acetate 3:1).

15: m.p. 78°C (hexane); $[\alpha]_D^{20} +9^\circ$ (c 1.93, chloroform); IR_{max}^{Nujol}: 1720, 1280, 1050 (ester), 1610, 1595 cm^{-1} (Ar). $^1\text{H NMR}$: δ 8.30-8.10 (m, 8H, arom.), 5.69 (m, 3-H), 5.19 (dd,

$J = J' = 10$, 4-H), 4.85 (d, $J = 3$, 1-H), 4.09 (m, 5-H), 2.48 (dd, $J = 13$, $J' = 5$, 2e-H), 2.02 (m, $J = 13$, $J' = 10$, $J'' = 3$, 2a-H), 1.30 (d, $J = 7$, 6-CH₃).

Anal. Calcd for C₂₁H₂₀N₂O₁₀ (460.41) : C, 54.78, H, 4.38 ; Found : C, 54.92, H, 4.29

16 : m.p. 164°C (hexane-acetone) ; $[\alpha]_D^{20} = -76^\circ$ (c 1.13, chloroform) ; IR : 3700 (OH), 1720, 1275, 1050, 1040 (ester), 1610 cm⁻¹ (Ar). ¹H NMR : δ 8.30 (d, $J = 9$) and 8.24 (d, $J = 9$) (AB, 4H, arom.), 5.33 (m, 3-H), 4.77 (d, $J = 3$, 1-H), 3.74 (m, 5-H), 3.41 (dd, $J = J' = 10$, 4-H), 3.33 (s, OCH₃), 2.31 (dd, $J = 13$, $J' = 5$, 2e-H), 1.89 (m, $J = 13$, $J' = 10$, $J'' = 3$, 2a-H), 1.36 (d, $J = 7$, 6-CH₃).

Anal. Calcd for C₁₄H₁₇NO₇ (311.30) : C, 54.01, H, 5.50, O, 35.97 ; Found : C, 54.15, H, 5.60 ; O, 36.01.

Methyl 2,6-dideoxy-3-O-methyl and methyl 2,6-dideoxy-4-O-methyl α -L-arabino-hexopyranosides (17 and 18). Treatment of 8 (500 mg, 3 mmol) according to procedure C gave 255 mg (52 %) of 17 and 100 mg (\approx 18 %) of 18 separated by column chromatography (hexane-ethyl acetate, 3:1). Physical and spectroscopic data were in agreement with the literature¹³ for 17 while 18 could be crystallized from hexane-acetone.

18 : m.p. 75-77°C, $[\alpha]_D^{20} = -105^\circ$; ¹H NMR : δ 4.68 (d, $J = 3$, 1-H), 3.90 (m, 3-H), 3.60-3.48 (m, 5-H), 3.54 (s, OMe), 3.28 (s, OMe), 2.67 (dd, $J = J' = 9$, 4-H), 2.07 (m, $J = 13$, $J' = 5$, 2e-H) and 1.65 (m, $J = 13$, $J' = 9$, $J'' = 3$, 2a-H), 1.32 (d, $J = 7$, 6-CH₃).

Anal. Calcd. for C₈H₁₆O₄ (176.21) : C, 54.53 H, 9.15, O, 36.32 ; Found : C, 54.65 H, 9.32 O, 36.25.

References and Notes

1. C. Monneret, J. Boivin, A. Martin and M. Pais, *Anthracycline Antibiotics* edited by H.S. El Khadem, Academic Press, New York, 1982, p. 25.
2. C. Monneret, A. Martin and M. Pais, *Tetrahedron Letters*, **27**, 575 (1986).
3. A. Martin, C. Monneret and M. Pais, *Carbohydr. Res.*, in press.

4. A. Martin, M. Pais and C. Monneret, Carbohydr. Res., **113**, 21, (1983).
5. D. Wagner, J.P.H. Verheyden and J.G. Moffatt, J. Org. Chem., **39**, 24 (1974).
6. See for example: S. David and S. Hanessian, Tetrahedron, **41**, 643 (1985).
7. T.-L. Su, R.S. Klein and J.J. Fox, J. Org. Chem., **47**, 1506 (1982) and references cited therein.
8. J. Alais and A. Veyrières, J. Chem. Soc., Perkin Trans I, 377, (1981).
9. T.M. Cheung, D. Horton and W. Weckerle, Carbohydr. Res., **74**, 93 (1979).
10. J.S. Brimacombe and K.M.M. Rahman, J. Chem. Soc. Perkin Trans I 1067 (1985).
11. J.P. Marsh, Jr., C.W. Mosher, E.M. Acton and L. Goodman, J. Chem. Soc. Chem. Comm., 973 (1967); G. Grethe, T. Mitt, T.H. Williams and M.R. Uskokovic, J. Org. Chem., **48**, 5309 (1983).
12. A.F. Hadfield, L. Cunningham and A.C. Sartorelli, Carbohydr. Res., **72**, 93 (1979).
13. K. Koga, S. Yamada, M. Yoh and T. Mizoguchi, Carbohydr. Res., **36**, C9 (1974); C. Monneret, C. Conreur and Q. Khuong-Huu, Ibid, **65**, 35 (1978).
14. S. David, A. Thiéffry and A. Forchioni, Tetrahedron Letters, **22**, 2647 (1981); C.W. Holzapfel, J.M. Koekemoer, C.F. Marais, G.J. Kruger and J.A. Pretorius, S. Afr. J. Chem., **35**, 80 (1982).
15. S.K. Gupta, Carbohydr. Res., **37**, 381 (1974); W.W. Lee, H.Y. Wu, J.E. Christensen, L. Goodman and D.W. Henry, J. Med. Chem., **18**, 768 (1975).
16. F. Staričskai, I. Pelyvas, R. Bognar and Gy. Butjas, Tetrahedron Letters, 1111 (1975); W.W. Lee, H.Y. Wu, J.P. Marsh, Jr., C.W. Mosher, E.M. Acton, L. Goodman and D.W. Henry, J. Med. Chem., **18**, 767 (1975); F. Sztaričskai, I. Pelyvas, L. Szilagyi, R. Bognar, J. Tamas and A. Neszmélyi, Carbohydr. Res., **65**, 193, (1978).
17. P. Bartner, D.L. Boxler, R. Brambilla, A.K. Mallams, J.B. Morton, P. Reichert, F.D. Sancilio, H. Surprenant, G. Tomalesky, G. Luckacs, A. Olesker, T.T. Thang, L. Valente and S. Omura, J. Chem. Soc., Perkin Trans I, 1600, (1979).